Computation of an Extractive Distillation Column with Affine Arithmetic

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The need of reliably solving systems of nonlinear equations often arises in the every-day practice of chemical engineering. In general, standard methods cannot provide theoretical guarantee for convergence to a solution, cannot reliably find multiple solutions, and cannot prove nonexistence of solutions. Interval methods provide tools to overcome these problems, thus achieving reliability. To the authors' best knowledge, computation of distillation columns with interval methods have not yet been considered in the literature. This article presents significant enhancements compared with a previously published interval method of the authors. The proposed branch-and-prune algorithm is guaranteed to converge, and is fairly general at the same time. If no solution exists then this information is provided by the method as a result. Power of the suggested method is demonstrated by solving, with guaranteed convergence, even the MESH equations of a 22 stage extractive distillation column with a ternary mixture.

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Introduction

Computing steady states of counter-current multistage processes is equivalent to finding solutions of large scale systems of nonlinear equations. Although a good deal of effort has been made in constructing efficient and robust computation techniques, and impressive results have been achieved, 1,2 generally there is no theoretical guarantee for

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convergence to the true solution. The routines developed for computing steady states are sensitive to initial estimates, and if no solution is achieved after several attempts with different initial points then one does not know whether the initial estimation is poor or simply that no solution exists for the specified circumstances. Moreover, there are specifications that give rise to several solutions (output multiplicity^{3,4}) but standard methods cannot guarantee that all solutions are found.

Interval methods provide tools to overcome these problems: these tools either provide all the solutions or prove nonexistence of solution of a general system of nonlinear

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equations with mathematical certainty. The Interval Newton/ Generalized Bisection method (IN/GB) has been successfully applied to solve a wide variety of chemical engineering problems⁵ such as computation of phase stability with activity coefficient models, ^{6,7} cubic equation-of-state (EOS) models, ^{8,9} modeling liquid–liquid equilibrium of ionic liquid systems, ¹⁰ calculation of critical points from cubic EOS models, ¹¹ location of azeotropes, ¹² parameter estimation using standard least squares and error-in-variables. ¹³ Interval arithmetic can also be applied to compute validated solutions of initial value problems for ODEs, ^{14,15} to enclose all solutions of two-point boundary value problems for ODEs,16 and to deterministic global optimization of nonlinear dynamic systems. 17

Interval methods improved considerably during the past few decades. State-of-the-art variants of IN/GB, involving advanced preconditioning, 18 linear programming 19 and/or constraint propagation on directed acyclic graphs (DAG),^{20–22} may be several orders of magnitude faster than the "textbook" Interval Newton/Gauss-Seidel²³ (IN/GS) algorithm with the so-called midpoint inverse preconditioner.

A new linearization technique, based on affine arithmetic (AA), ^{24–27} has been proposed recently by Kolev. ^{28–36} Numerical evidence published in the literature ^{28–32,34,37–41} suggests that the new technique may be superior to the traditional linearization techniques such as the interval Newton or the Krawczyk⁴² method. Linear programming may be preferable as pruning technique for this new linearization in the case of the vapor-liquid equilibrium cascades.⁴¹

The aim of the present work is to combine the aforementioned ideas to obtain an efficient interval methodology, and thus extend the capability of these methods to compute such a complex and large scale chemical engineering problem as the steady state of an industrial scale distillation column. Power of the suggested method is demonstrated by solving a 22 stage extractive distillation column with a ternary mixture. To the authors' best knowledge, computation of industrial scale distillation columns with interval methods have not yet been considered in the literature. It is perhaps so because of the extensive complexity and dimensionality of these problems. The proposed method is also able to provide information on infeasibility if the system of equations has no solution, and is able to find several solutions in the studied domain if they exist.

Procedure for Locating All Solutions

Here the procedure used in this work for locating all the solutions is described. Three major components of the procedure may be distinguished: linearization, pruning (discarding some regions of the variables' domain not containing a solution), and bisection.

Linearization

Given

$$\mathbf{f}(\mathbf{x}) = \mathbf{0}$$
, where $\mathbf{f} : \mathbb{R}^n \to \mathbb{R}^n$, $x_i \in X_i = [\underline{x}_i, \overline{x}_i]$ (1)

The goal is to bound all solutions of (1) or prove their absence using a first order interval method. Linearization of (1) with the mixed affine arithmetic and interval arithmetic model²⁵ (mixed AA/IA, pp. 75–76) yields a linear constraint system in the form of

$$A(X)\mathbf{x} + B(X) = \mathbf{0} \quad \mathbf{x} \in X \tag{2}$$

which must be satisfied by any of the solution vector(s) $\mathbf{x}^* \in$ X; where A(X) is a real $n \times n$ matrix and B(X) is an interval vector.

The mixed AA/IA was used only at the critical parts (where otherwise division by zero or calling the logarithm function with negative argument would have occurred) in the previous work⁴¹ of the authors due to implementation design flaws. Based on the conclusions of the previous work, the affine class has been redesigned and implemented in C++. The mixed AA/IA is used during the entire solution process in the present work. All the optimization techniques proposed in the monograph²⁵ are incorporated (pp. 79–83); most noticeably the affine class uses a memory pool which is automatically managed by the constructors and destructors of the affine class.

Pruning based on constraint propagation

Two methods are used for discarding from the box some regions not containing a solution (shortly: for pruning). One of them is based on equationwise constraint propagation³²: the formula

$$X_j^{\text{new}} = X_j \cap \left(\frac{1}{a_{ij}} \left(B_i - \sum_{k \neq j} a_{ik} X_k\right)\right)$$
(3)

is evaluated equation by equation, and for each variable in the actual equation. In formula (3), a_{ij} , X_i , and B_i are the corresponding elements of the real matrix A, interval vector X and B in (2), respectively. Redundant equations can also be involved in the aforementioned propagation.

Equation (3) is the affine analogue of the well-known Interval Newton Gauss-Seidel iteration: in (3) the denominator is real, whereas in the Gauss-Seidel iteration division by an interval is required.

Pruning based on linear programming

The other method of pruning is based on linear programming; this is the so-called LP pruning. The original nonlinear function (1) is enclosed by the linear enclosure (2). Tight bounds on the solution set of (2), hence on the solution set of (1), is computed by solving the linear programming subproblems below:

$$\min/\max x_j \text{ for all } j$$
subject to
$$A\mathbf{x} = -\mathbf{B}$$

$$\mathbf{x} \in \mathbf{X}$$
(4)

where the constraints are the same as (2) and remain unchanged during the entire pruning procedure.

At first glance it seems as if 2n LP subproblems have to be solved (n denotes the number of variables) but this is not the case. The minimization/maximization subproblem for x_i can be skipped if x_i equals its lower/upper bound, respectively, in any of the primal feasible solution vectors obtained during the pruning procedure. The gain is obvious.

Any primal feasible solution of the LP subproblems (4) remains primal feasible after manipulating the objective arbitrarily. It follows that only the first LP subproblem has to be solved from scratch; all other subproblems should use the optimal solution of the preceding subproblem as a primal feasible basis and run only Phase II of the primal simplex algorithm, thus hopefully reducing the computational efforts. The naïve sequence⁴¹ to process the x_i variables would be $\min x_1$, $\max x_1$, $\min x_2$, $\max x_2$, ... etc but the subproblems $\min x_1$ and $\max x_1$ are likely to produce completely different solutions. As a consequence, this is expected to result in a lot of simplex iterations when using the optimal solution of subproblem min x_1 as the initial primal feasible basic solution when solving the subproblem max x_1 .

Considering this idea, a simple heuristic is proposed for selecting the subsequent subproblem. Find that variable which is the closest to its lower/upper bound and has not yet been considered in the pruning step; then solve the corresponding optimization problem (min x_i if x_i is close to its lower bound, or max x_i if x_i is close to its upper bound). The assumption behind this heuristic is that the current primal feasible solution should not be far from the optimal solution for that variable. The enhancements presented in this subsection will be referred to as Achterberg's heuristic. Numerical examples, suggesting the superiority of this heuristic to the previous implementation, 41 will be presented after subsection Separation problem which describes the numerical examples used for comparison.

Bisection

A simple rule is used in this article: bisect the box along the domain of the widest component. If the problem is badly scaled globally, it is desirable to choose the scale factors after the first LP pruning step so as all edges of the box equal unity. Unfortunately, this bisection rule may not be robust enough in general as it is demonstrated at the Numerical examples section. The choice of the variable along which to bisect the box is a hard and open question in general. In the case of the Separation problem section below, however, a simple yet efficient problem specific bisection rule can be constructed if the above simple rule fails; numerical examples are also presented at subsection Problem-specific bisection rule.

Branch-and-prune algorithm to bound all solutions

Step 0: Initialize a stack of boxes with the original box.

Step 1: If the stack is empty then exit else pop the topmost box $X^{(k)}$ off the stack.

Step 2: Linearize the system of equations in $X^{(k)}$ with the mixed affine arithmetic and interval arithmetic model. If the obtained lower and upper bounds on the range of f do not enclose 0 then discard the box and go to Step 1.

Step 3: Apply equationwise constraint propagation; if an empty interval is obtained then discard the box and go to Step 1.

Step 4: Relinearize f in the (hopefully) contracted box, and apply LP pruning. If the first LP problem is infeasible then discard the box and go to Step 1.

Step 5: If the widest component of the contracted box is below a predefined threshold then print the box (it may contain a solution) else bisect it along the domain of the widest component, push the resulting two boxes to the stack, and go to Step 1.

Note: Steps 2, 3, and 4 may be repeated if the box is sufficiently reduced in size (analogous to the idea of Hansen, ⁴³ pp. 98-100). However, it is not straightforward how to quantify that the box is "sufficiently" contracted. Instead, one can simply repeat steps 2, 3, and 4 for a fixed number of times irrespective of the rate of pruning; this is obviously not the most effective way but is easy to implement.

Effect of the Enhancements Concerning the Linearization

As discussed earlier at subsection Linearization, the present implementation uses the mixed affine arithmetic/interval arithmetic (mixed AA/IA) model during the entire computations, and the affine class uses a memory pool. The expected effects of these enhancements are first drawn up, and then comparisons to the previous results⁴¹ are given. The numerical results confirm the expectations.

Using the Chebyshev approximation (Stolfi and Figueiredo,²⁵ pp. 56–57), the difference between the mixed AA/IA and the pure affine arithmetic is that the so-called overshoot (p. 63) is cut off in the mixed AA/IA leading to better approximation. Note that the overshoot is unwanted because it can result in division by zero, or calling the logarithm function with negative argument.⁴¹ The mixed AA/IA with Chebysev approximation gives identical results with respect to the range of the enclosure as the pure affine arithmetic with min-range approximation (pp. 64-65), while the enclosure of the mixed AA/IA is mostly tighter than the pure affine arithmetic with min-range approximation. See also Figures 1a-d. The better/mostly better approximation is expected to give smaller number of iterations.

The previous implementation was based on the map container of the C++ Standard Template Library, the present one uses arrays arranged in a memory pool. This enhancement obviously cannot influence the number of iterations but is expected to give at least an order of magnitude speed-up (Item 10 in the book of Meyers⁴⁴). The numerical examples with the AA/CP method of Baharev and Rév⁴¹ are recomputed; these examples do not involve linear programming which is important to maintain comparability since the LP pruning part is also changed in the current implementation. The software and hardware environment is given in the appendix. As shown in Tables 1-3, the numerical examples confirm the expectations.

Liquid phase split

Given the overall composition, the goal is to determine the relative amounts and compositions of at most two phases in equilibrium. The equifugacity conditions are solved at constant pressure and temperature.

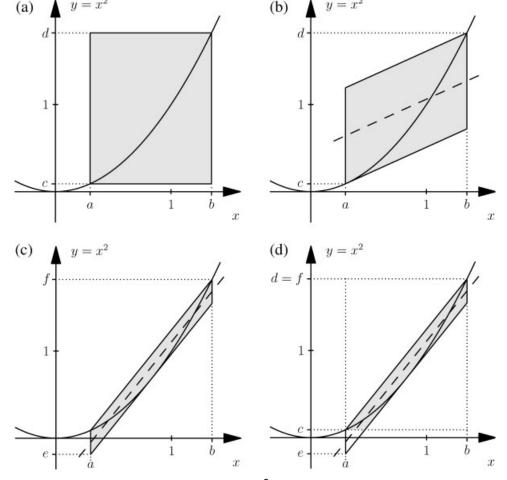


Figure 1. Illustration of the linear enclosure of function x^2 computed by (a) ordinary interval arithmetic (IA), (b) affine arithmetic (AA) and min-range approximation, (c) AA and Chebyshev approximation, (d) mixed AA/IA model with Chebyshev approximation.

The slope of the dashed line computed by AA correlates well with the slope of the approximated function.

Counter-current equilibrium cascade

The steady state of one theoretical stage between a reboiler and a condenser (VLE cascade with three stages) is computed by solving the MESH equations (component material balances, equilibrium conditions, summation equations, and heat balance equations) simultaneously.

Separation Problem

Steady state of continuous extractive distillation of acetone and methanol with water as entrainer is computed. The equilibrium stage used in this work is shown in Figure 2; the scheme of the studied distillation column is given in Figure 3.

Table 1. Comparison of the Previous and Current Implementation (LLE Phase Split, Binary Mixture)

	Impleme	ntation	
	Previous ⁴¹	Current	Previous/Current
Time (s)	1.15	0.010	115
Iterations	1407	627	2.24
Cycle time (μ s)	817	15.9	51.3

Variables and specifications

Specifications are the reflux ratio R, distillate flow rate D, composition and flow rate of the solvent feed and the main feed, total number of stages, and feed stage locations. As shown in Figure 3, total condenser and total reboiler is used. Variables are listed in Table 4.

Enthalpy model

A fairly simplified enthalpy model is used: the molar enthalpy of the vapor phase is the mole fraction weighted average of the constant molar heat of vaporization (λ_i) of the components, $H = \Sigma \lambda_i y_i$. Other heat effects are neglected.

Table 2. Comparison of the Previous and Current Implementation (LLE Phase Split, Ternary Mixture)

	Impleme		
	Previous ⁴¹	Current	Previous/Current
Time (s)	23.3	0.790	29.5
Iterations	7715	6513	1.18
Cycle time (μ s)	3010	121	24.9

Table 3. Comparison of the Previous and Current Implementation (VLE Cascade with Three Stages)

	Impleme		
	Previous ⁴¹	Current	Previous/Current
Time (s)	4.44	0.099	44.8
Iterations	1687	645	2.62
Cycle time (μ s)	2630	153	17.1

The molar heat of vaporization values are given in Table A1. The assumption behind this model is that the heat of vaporization is at least an order of magnitude higher than the other enthalpy changes in the liquid or vapor phase, which is reasonable for distillation in practice. This model may seem rough but when comparing the computational results performed with this model to those obtained with commercial simulators using detailed and thermodynamically consistent enthalpy model, the result are the same up to 2–3 digits.

Equations related to the variables involved in the pruning

These are the so-called MESH equations. Component material balance (M) equations:

$$l_{ij-1} + v_{ij+1} + f_{ij} - (l_{ij} + v_{ij}) = 0$$
 $i = 1...C;$ $j = 1...N$
 $l_{ij} = L_j x_{ij}$ $i = 1...C;$ $j = 0...N$
 $v_{ij} = V_j y_{ij}$ $i = 1...C;$ $j = 1...N + 1$

where $f_{i,j}$ is the specified molar flow rate of component i in feed stream to stage j.

Vapor-liquid equilibrium (E) equations:

$$y_{ii} = K_{ii}x_{ii}$$
 $i = 1...C$; $j = 1...N$

Summation (S) equations:

$$\Sigma x_{ij} = 1$$
 $j = 1...N$

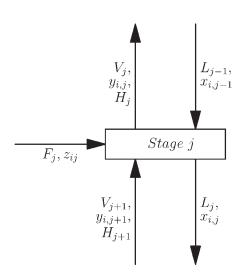


Figure 2. Equilibrium stage.

$$\Sigma y_{ij} = 1$$
 $j = 1...N$

Heat balance (H) equations:

$$Q_j = Q_{j+1} \quad j = 1...N$$

Auxiliary equations:

$$H_{j} = \sum \lambda_{ij} y_{ij} \quad j = 2...N + 1$$

$$Q_{j} = V_{j} H_{j} \quad j = 2...N + 1$$

$$K_{ij} = \gamma_{ij} p_{ij} / P \quad i = 1...C; \quad j = 1...N$$
(5)

When computing the linearized form of $K_{i,j}$ according to Eq. 5 with the mixed AA/IA model, both for equationwise constraint propagation and linear programming-based pruning, all noise variables (basic entities of the AA model, see Ref. 25, pp. 43–44) not corresponding to \mathbf{x} or T are combined into a single noise variable (Ref. 25, pp. 81–82). This is obviously a loss of information but it makes the structure of the LP problem simpler and improves the scaling.

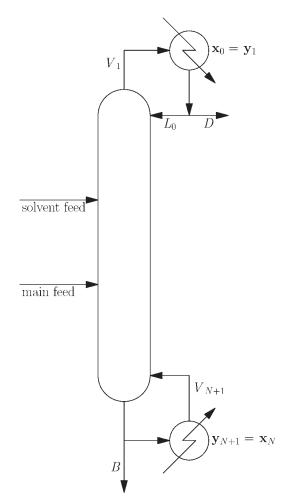


Figure 3. Continuous extractive distillation.

Table 4. List of Variables*

		Involved in Pruning	Not Involved in Pruning
$x_{i,j}$	Mole fraction of component i in the liquid phase at stage j	$j=1\ldotsN$	j = 0
$y_{i,j}$	Mole fraction of component i in the vapor phase at stage j	$j=1\ldotsN$	j = N + 1
$egin{array}{c} y_{i,j} \ K_{i,j} \end{array}$	Equilibrium ratio	$j=1 \ldots N$	
$l_{i,j}$	Molar flow rate of component i in the liquid phase at stage j	j = 1 N	j = 0
$v_{i,j}$	Molar flow rate of component i in the vapor phase at stage j	j = 2 N + 1	j = 1
$V_i^{"}$	Vapor flow rate at stage <i>j</i>	j = 2 N + 1	j = 1
H_i	Molar enthalpy of vapor at stage j	j = 2 N + 1	j = 1
$\vec{Q_i}$	Enthalpy rate carried by the vapor at stage j	j = 2 N + 1	j = 1
T_{j}	Temperature at stage j	$j=1 \dots N$	

Stage j = 0 is the condenser, stage j = N+1 is the reboiler.

Vapor-liquid equilibrium conditions are modeled with a modified Raoult-Dalton equation. Liquid phase activity coefficients are modeled by the 2-parameter Wilson equations

$$\ln \gamma_{i} = -\ln \left(\sum_{a=1}^{C} x_{a} \Lambda_{ia} \right) + 1 - \sum_{b=1}^{C} \frac{x_{b} \Lambda_{bi}}{\sum_{a=1}^{C} x_{a} \Lambda_{ia}} \quad i = 1...C$$
(6)

$$\Lambda_{ab} = \frac{V_b^m}{V_a^m} \exp\left(-\frac{k_{ab}}{R_G T}\right) \quad a = 1...C; \quad b = 1...C \quad (7)$$

where model parameters k_{ab} and V_i^m are given in Tables A2 and A3, respectively; R_G is the general (Regnault's) gas constant, given at Table A2; T is the temperature; pure components vapor pressures p_i are computed by the Antoine equation

$$ln p_i = A_i - \frac{B_i}{C_i + T}$$
(8)

with coefficients A_i , B_i , C_i given in Table A4; pressure P at each stage is specified to be 101,325 Pa, i.e., no pressure drop is taken into account for simplicity.

Note that Eqs. 6-8 are not directly involved in the pruning but only the linearized equation (Eq. 5) is.

Equations related to variables not involved in the pruning

These variables are computed by substitution to the following equations:

At stage j = 0 (total condenser):

$$x_{i,0} \leftarrow y_{i,1}$$
 $i = 1...C$

$$l_{i,0} \leftarrow RDy_{i,1}$$
 $i = 1...C$

At stage j = 1 (upmost equilibrium stage):

$$v_{i,1} \leftarrow (R+1)Dy_{i,1} \quad i = 1...C$$

$$V_1 \leftarrow (R+1)D$$

$$H_1 \leftarrow \sum \lambda_i y_{i,1}$$

$$O_1 \leftarrow O_2$$

At stage j = N + 1 (reboiler):

$$y_{i,N+1} \leftarrow x_{i,N}$$
 $i = 1...C$

Numerical Examples

Specifications

Components are (1) acetone, (2) methanol, and (3) water (C = 3). Specifications are R = 5, D = 0.73 mol/s, solvent (entrainer) feed is 2.0 mol/s pure water, main feed is 0.783 mol/s acetone and 0.217 mol/s methanol. The location of each feed tray is specified. The column has N stages, plus a total condenser and a total reboiler (stage 0 and stage N + 1, respectively).

Purity restriction

The purity restriction on the mol fraction of the acetone in the distillate is varied; three cases are discussed: $0.96 \le$ x_{acetone} , $0.92 \le x_{\text{acetone}}$, $0.78 \le x_{\text{acetone}}$. Note that the last case is included barely for testing numerically the proposed method; it is too permissive and is not meaningful from

Table 5. Computational Results for the Extractive Distillation Column with Different Specifications and Restrictions[§]

Number of Trays	Purity Restriction on Distillate	Distillate Composition	Time (s)	Number of Boxes	Simplex Iterations
12*	$0.96 \le x_{\text{acetone}}$	Infeasible	2.77	3	23,891
12	$0.92 \le x_{\text{acetone}}$	(0.923, 0.0430, 0.0342)	22.10	19	247,522
12	$0.78 \le x_{\text{acetone}}$	Time limit reached	>12,000	-	-
16^{\dagger}	$0.96 \le x_{\text{acetone}}$	Infeasible	9.49	9	78,803
16	$0.92 \le x_{\text{acetone}}$	(0.942, 0.0343, 0.0234)	54.15	29	500,041
16	$0.78 \le x_{\text{acetone}}$	Time limit reached	>12,000	_	_
22^{\ddagger}	$0.96 \le x_{\text{acetone}}$	(0.961, 0.0212, 0.0179)	52.86	15	459,290
22	$0.92 \le x_{\text{acetone}}$	(0.961, 0.0212, 0.0179)	92.13	33	709,058
22	$0.78 \le x_{\text{acetone}}$	Time limit reached	>12,000	-	-

^{*}Solvent feed to tray 5, main feed to tray 9.

^{*}The Domain of i is $1 \dots C$ in all cases

Solvent feed to tray 7, main feed to tray 12. *Solvent feed to tray 9, main feed to tray 16.

^{§(}General Splitting Rule is Applied)

engineering aspect because distillate stream significantly richer in acetone than the azeotropic composition is to be produced in this operation.

Preparation of the initial box

The initial intervals of V_j are obtained by a reasonable assumption from engineering aspect: $abs((V_j - V_1)/V_1) \le 0.32$ where $V_1 = (R+1)$ D=4.38 mol/s as it follows from the specifications. Note that the constant molar overflow assumption would involve $abs((V_j - V_1)/V_1) = 0.0$.

The initial intervals for all the $K_{i,j}$ values are chosen to be $K_{1,j} \in [0.98, 38.97], K_{2,j} \in [0.80, 7.53], \text{ and } K_{3,j} \in [0.26, 1.5]$ 1.01]; these properly enclose the range of *K* in the $\mathbf{x} \in [0,1]^C$ space. These intervals of the $K_{i,j}$ values are read from the graph of the corresponding K_i implicit function over $\mathbf{x} \in$ [0,1]^C. These values are then verified by proving the infeasibility of the corresponding system of equations with the proposed method. For example to verify the lower bound 0.98 on K_1 , the interval equation $K_1 = [-\infty, 0.98]$ is appended to the system of equations describing the bubble point of the mixture, and then its infeasibility is proven by the proposed branch-and-prune method. Verifying all the bounds similarly totals less than a second. The initial T_i values are [327,374]; this interval properly encloses all possible bubble points at the specified pressure. All mole fractions are assumed to be not less than 0.01. All other initial intervals are chosen to be noninformative, i.e., $[-\infty, +\infty]$.

Results

The results are given in Table 5. Example profiles are shown in Figures 4 and 5. The restriction $0.96 \le x_{\rm acetone}$ proves to be too strict in case N=12 and N=16, the separation problem is infeasible. This information is provided by

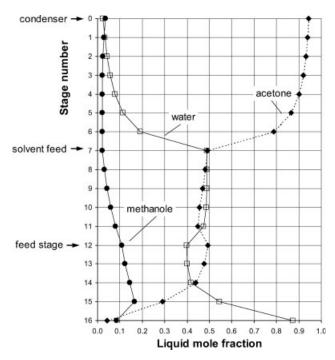


Figure 4. Composition profile.

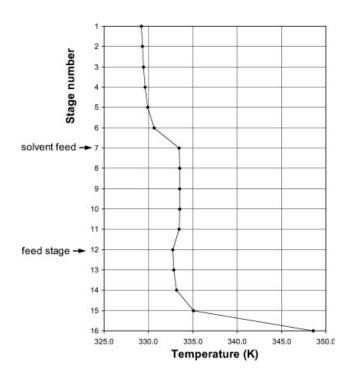


Figure 5. Temperature profile.

the method as a result. The problem becomes feasible with the more permissive but still reasonable $0.92 \le x_{\rm acetone}$ constraint, and the solution is successfully found. In the practically meaningless case of the $0.78 \le x_{\rm acetone}$, used here for numerical testing only, the solution cannot be found in 3 h with the machine used for computation (software and hardware environment is given in the appendix, the used LP solver is GNU GLPK 4.28⁴⁵). Note, however, that a general splitting rule is applied here, and the problem can be solved in a few minutes with a problem specific splitting rule, as discussed in the next subsection.

Problem-specific bisection rule

Splitting the box along its widest component proved to be unsatisfactory in case of the rather loose and practically meaningless restriction $0.78 \le x_{\rm acctone}$. The problem can be solved in minutes by multiplying the mol fraction of the acctone in the distillate by an appropriate weight w when applying the bisection rule. This is shown in Table 6. If the multiplier is too big, e.g., 50 or larger in our case, then always the interval being multiplied by w is bisected which is obviously not the ideal solution, hence the slightly worse results. However, the problem is solved in minutes even in that case.

Table 6. Effect of Changing the Multiplier w Used in the Bisection Step

Multiplier w	Time(s)	Number of Boxes	Simplex Iterations
1	>12,000	_	_
10	>12,000	_	_
20	157.5	99	1,424,169
≥50	187.8	113	1,787,940

Specifications: Number of Trays: 16, Solvent Feed to Tray: 7, Main Feed to Tray: 12, Purity Restriction on the Distillate: $0.78 \le x_{\rm acctone}$.

Table 7. Comparison of the Computational Results Using Multiplier w = 1 and w = 20 in the Bisection Step

Number of Trays	Purity Restriction on Distillate	Time (s) $(w = 1)$	Time (s) $(w = 20)$	Number of Boxes $(w = 1)$	Number of Boxes ($w = 20$)	Simplex Iterations $(w = 1)$	Simplex Iterations $(w = 20)$	$\frac{\mathrm{Time}_{w=1}}{\mathrm{Time}_{w=20}}$	$\frac{\text{Simp. iter}_{w=1}}{\text{Simp. iter}_{w=20}}$
12*	$0.96 \le x_{\text{acetone}}$	2.77	4.19	3	7	23,891	40,601	0.66	0.59
12	$0.92 \le x_{\text{acetone}}$	22.10	17.37	19	17	247,522	200,165	1.27	1.24
12	$0.78 \le x_{\text{acetone}}$	>12,000	67.50	_	67	_	759,990	>177.78	_
16^{\dagger}	$0.96 \le x_{\text{acetone}}$	9.49	16.48	9	13	78,803	152,683	0.58	0.52
16	$0.92 \le x_{\text{acetone}}$	54.15	56.51	29	31	500,041	577,726	0.96	0.87
16	$0.78 \le x_{\text{acetone}}$	>12,000	157.5	_	99	_	1,424,169	>76.19	_
22^{\ddagger}	$0.96 \le x_{\text{acetone}}$	52.86	210.94	15	55	459,290	1,934,330	0.25	0.24
22	$0.92 \le x_{\text{acetone}}$	92.13	230.26	33	61	709,058	2,029,564	0.40	0.35
22	$0.78 \le x_{acetone}$	>12,000	498.03	_	147	-	3,968,612	>24.09	_

^{*}Solvent feed to tray 5, main feed to tray 9.

This problem specific rule generally produces worse results for the reasonable purity restrictions, as it is shown in Table 7, but the computation time remains acceptable. Note that the original bisection rule corresponds to the case w = 1.

Effect of the enhancements concerning the LP pruning

The Achterberg's heuristic presented at subsection Linear programming-based pruning is expected to reduce the number of simplex iterations, thus the overall computation time. The numerical examples of Baharev and Rév⁴¹ can be solved in less then a second with the current implementation, as it was presented earlier in Tables 1-3, which makes those examples unsuitable for testing. The Separation problem is chosen instead. As shown in Table 8, the Achterberg's heuristic makes the pruning roughly 5 times faster, which is in line with the reduced number of simplex iterations.

Finding multiple solutions

The ability of the proposed method to prove nonexistence of solutions is already demonstrated earlier. The capability of finding multiple solutions is presented here. As the separation problem has a single solution, the test problem of Meintjes and Morgan, 46 chemical equilibrium of hydrocarbon combustion, is computed instead. All the four solutions are found; they are shown in Table 9. This requires 1.46 seconds and 22,219 boxes to be examined.

Variables

$$x_1 \dots x_5$$
 all in [-1.0 E +1, 1.0 E +8],

Table 8. Comparison of the Computational Results Obtained with/without Using Achterberg's Heuristic

	Without Heur.	With heur.	Without Heur./ With Heur.
$0.96 \le x_{\text{acetone}} \text{ and } w = 1$			
Time (s)	44.0	9.49	4.63
Simplex iterations	505,424	78,803	6.41
$0.92 \le x_{\text{acetone}}$ and $w = 1$			
Time (s)	301.43	54.15	5.57
Simplex iterations	3,376,560	500,041	6.75
$0.78 \le x_{\text{acetone}}$ and $w = 20$			
Time (s)	769.56	157.5	4.89
Simplex iterations	9,102,883	1,424,169	6.39

Number of trays is 16.

Constants

R = 10,

 $R_5 = 0.193$,

 $R_6 = 0.002597/\sqrt{40},$

 $R_7 = 0.003448/\sqrt{40}$

 $R_8 = 0.00001799/40,$

 $R_9 = 0.0002155/\sqrt{40},$

 $R_{10} = 0.00003846/40;$

Equations

$$3x_5 = x_1(x_2 + 1),$$

$$x_3(x_2(2x_3 + R_7) + 2R_5x_3 + R_6) = 8x_5,$$

$$x_4(R_9x_2 + 2x_4) = 4Rx_5,$$

$$x_2(2x_1 + x_3(x_3 + R_7) + R_8 + 2R_{10}x_2 + R_9x_4) + x_1 = Rx_5,$$

$$x_2(x_1 + R_{10}x_2 + x_3(x_3 + R_7) + R_8 + R_9x_4) + x_1$$

$$x_3(R_5x_3 + R_6) + x_4^2 = 1.$$

Limitations of the current implementation

The proposed method is fairly general, and the numerical results are promising. However, the implementation is still in its infancy. The C++ source code of the affine class consists of ~3000 lines although only the bare minimum of the functions are implemented. The C++ code of the distillation column is ~2300 lines. The source code was developed solely for experimental purposes, i.e., to study the numerical capabilities of the proposed algorithm, thus it is hard to extend or modify. Interfacing the solver with a modeling language would make the work drastically easier. Debugging of the C++ source code is difficult since the authors do not know any mixed affine arithmetic/interval arithmetic implementation that could give correct reference values in case of a suspected bug.

Table 9. All Real Solutions of the Chemical Equilibrium of Hydrocarbon Combustion Problem in the Given Search Box

	Solution 1	Solution 2	Solution 3	Solution 4
x_1	3.114 E-3	2.757 E-3	2.471 E-3	2.153 E-3
x_2	3.460 E+1	3.924 E+1	4.388 E+1	5.055 E+1
x_3	6.504 E-2	-6.139 E-2	5.778 E-2	-5.414 E-2
x_4	8.594 E-1	8.597 E-1	-8.602 E-1	-8.607 E-1
x_5	3.695 E-2	3.699 E-2	3.697 E-2	3.700 E-2

[†]Solvent feed to tray 7, main feed to tray 12. ‡Solvent feed to tray 9, main feed to tray 16.

Summary

Generally, there is no theoretical guarantee for convergence to the true solution at computing steady states of counter-current multistage processes with conventional methods. The routines are usually sensitive to initial estimates, and if no solution is achieved after several attempts with different initial points then one does not know whether the initial estimation is poor or simply that no solution exists for the specified circumstances. Moreover, there are specifications that give rise to several solutions but standard methods cannot guarantee that all solutions are found.

Interval methods provide tools to overcome these problems: these tools either provide all the solutions or prove nonexistence of solution of a general system of nonlinear equations with mathematical certainty. This article presents significant enhancements compared to a previously published interval method⁴¹ of the authors: both the linearization and the linear programming-based pruning step are revised. The effect of each enhancement is demonstrated on the corresponding numerical examples of the previous work, namely the Liquid-phase split section with binary and ternary mixtures, and the Counter-current equilibrium cascade section with one theoretical stage is recomputed.

The aforementioned improvements make it possible to compute industrial scale distillation columns: MESH equations of columns, number of theoretical stages varying from 12 to 22, hosting continuous extractive distillation of acetone and methanol with water as entrainer are successfully solved. The authors consider this as the main achievement of the article: to the authors' best knowledge, computation of distillation columns with interval methods has not yet been considered in the literature.

Further numerical examples presented in the article lead to the following observations. The computation with the simple "split the widest interval" bisection rule is rather slow if unreasonably loose product purity is specified for the distillation column, but can be shortened to a few minutes by applying a problem specific weight factor in the bisection step. Applicability to proving nonexistence of solutions is also demonstrated, in case of specifications that cannot be met. Capability of the method to find multiple solutions is illustrated on a problem of chemical equilibria.

The proposed method is fairly general, and the numerical results are promising. The implementation is, however, still in its infancy, and it is very difficult to code the system of nonlinear equations in C++ programming language. The authors plan to hook the solver to an appropriate modeling language to make the usage of the solver easier.

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Appendix A

Table A1. Heat of Vaporization at Normal Boiling Point

	λ	-i
Component i	cal/mol	kJ/mol
Acetone	6960	29.12
Methanol	8426	35.25
Water	9717	40.66

Data is from the databank of ChemCAD 5.5.4 (Chemstations). Computations were carried out with the values in cal/mol.

Table A2. Parameters of the Wilson Equation

i	j	k _{ij} (cal/mol)	k _{ji} (cal/mol)	$K_{ij}/R_{\rm G}$ (K)	$K_{ji}/R_{\rm G}$ (K)
1	2	-157.981	592.638	79.4989	298.226
1	3	393.27	1430.0	197.90	719.60
2	3	-52.605	620.63	-26.472	312.31

Data is from the databank of ChemCAD 5.5.4 (Chemstations) $R_{\rm G}$ is the general (Regnault's) gas constant 8.314472 J/(mol K). Computations were carried out with the values in cal/mol and with $R_{\rm G}=1.98721$ cal/(mol K). Components: (1) acetone, (2) methanol, (3) water.

Table A3. Liquid Molar Volume

Component i	$V_i^m \text{ (cm}^3\text{/mol)}$
Acetone	74.05
Methanol	40.729
Water	18.069

Data is from the databank of ChemCAD 5.5.4 (Chemstations).

Table A4. Parameters of the Antoine Equation $\ln p = A - B/(T + C)$, where Vapor Pressure p is in mmHg, Temperature T is in Degrees Kelvin

Component i	A _i (for mmHg)	B_i (K)	$C_i(K)$	A _i * (for Pa)
Acetone	16.732	2975.9	-34.523	21.625
Methanol	18.51	3593.4	-35.225	23.40
Water	18.304	3816.4	-46.13	23.197

Data is from the databank of ChemCAD 5.5.4 (Chemstations). Computations were carried out with the A, B, C values; to get the vapor pressure in Pa use A^* instead of A.

Software and hardware environment

The computations are carried out with the following hardware and software configuration. Processor: Intel Pentium 4 530 Prescott at 3.00 GHz, L1 cache 16 KB, L2 cache 1024 KB; memory: 2 × 512 MB PC3200 DDR RAM 400 MHz (dual channel interleaved), bus speed 800 MHz; chipset: Intel i915P; operating system: Kubuntu 5.04 (in text mode) with Ubuntu kernel 2.6.10-5-686-smp; compiler: Intel C++ Compiler for Linux 8.1, compiler flags: -O2 -ip -static -xP.

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